Singlet Oxygen Toxicity Is Cell Line-dependent: A Study of Lipid Peroxidation in Nine Leukemia Cell Lines

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ABSTRACT

Singlet oxygen (1 O2) can be quenched by water, lipids, proteins, nucleic acids and other small molecules. Polyunsaturated fatty acids (PUFA) of cells principally quench 1 O2 by chemical mechanisms, producing lipid hydroperoxides, while proteins physically and chemically quench 1 O2. Because cell lines can have different PUFA and protein levels, we hypothesized that 1 O2 toxicity will vary between cell lines. We used Photofrin® as a source of 1 O2. Exposure of nine different leukemia cell lines (CEM, HEL, HL-60, K-562, KG-1, L1210, Molt-4, THP-1 and U-937) to Photofrin and light results in changes in membrane permeability (trypan blue) that vary with cell line. The greater the lipid content of the cell line, the less susceptible they are to membrane damage. When the cell media was supplemented with docosahexaenoic acid (DHA, 22:6), the overall unsaturation of cellular lipids increased. Photofrin and light resulted in increased radical formation in these supplemented cells compared to controls; however, there was no difference in membrane permeability between DHA-supplemented and control cells. Lipid-derived radical formation (electron paramagnetic resonance spin trapping) was cell line dependent; but no correlation between lipid content of cells and radical formation was found. However, we found that the greater the protein content of cells the more they were protected against membrane damage induced by Photofrin photosensitization. This suggests that cellular proteins are a key target for 1 O2-mediated toxicity. A remarkable observation is that cell size correlates inversely with ability of cells to cope with a given flux of 1 O2.

MATERIALS AND METHODS

Chemicals. Photofrin (poriferous sodium) was kindly provided by QLT Phototherapeutics, Inc., Vancouver, BC, Canada. It was dissolved in 5% dextrose (pH 7.4) and frozen at −20°C.

ABSTRACT

Singlet oxygen (1 O2) can be quenched by water, lipids, proteins, nucleic acids and other small molecules. Polyunsaturated fatty acids (PUFA) of cells principally quench 1 O2 by chemical mechanisms, producing lipid hydroperoxides, while proteins physically and chemically quench 1 O2. Because cell lines can have different PUFA and protein levels, we hypothesized that 1 O2 toxicity will vary between cell lines. We used Photofrin® as a source of 1 O2. Exposure of nine different leukemia cell lines (CEM, HEL, HL-60, K-562, KG-1, L1210, Molt-4, THP-1 and U-937) to Photofrin and light results in changes in membrane permeability (trypan blue) that vary with cell line. The greater the lipid content of the cell line, the less susceptible they are to membrane damage. When the cell media was supplemented with docosahexaenoic acid (DHA, 22:6), the overall unsaturation of cellular lipids increased. Photofrin and light resulted in increased radical formation in these supplemented cells compared to controls; however, there was no difference in membrane permeability between DHA-supplemented and control cells. Lipid-derived radical formation (electron paramagnetic resonance spin trapping) was cell line dependent; but no correlation between lipid content of cells and radical formation was found. However, we found that the greater the protein content of cells the more they were protected against membrane damage induced by Photofrin photosensitization. This suggests that cellular proteins are a key target for 1 O2-mediated toxicity. A remarkable observation is that cell size correlates inversely with ability of cells to cope with a given flux of 1 O2.

INTRODUCTION

Singlet oxygen (1 O2)† generated in biological systems can serve as a signaling molecule as well as an initiator of detrimental biological oxidations (1). It can be produced by photosensitization with Photofrin®. Photofrin is a clinically approved photosensitizer used in the photodynamic therapy (PDT) of various types of cancer. It is a hydrophobic, hematoporphyrin derivative that locates in cell membranes, where, upon light exposure, it produces singlet oxygen (2,3). Singlet oxygen reacts with polyunsaturated fatty acids (PUFA) in the cell membrane to form lipid hydroperoxides (4). Ferrous iron can act as a catalyst and initiate the breakdown of lipid hydroperoxides via a radical chain reaction. This chain reaction produces reactive oxygen species, aldehydes and other toxic products that can lead to cell death (5–7).

Both the protein and PUFA content of different cell lines can vary greatly. Various amino acids can quench 1 O2 physically; e.g. histidine has a rate constant of (k = 107−109 M−1 s−1) (8); however tryptophan, methionine and cysteine chemically quench 1 O2 resulting in protein damage (9–11). It has been shown that proteins can dominate the quenching of 1 O2 in the membrane and the cytosol (12,13). However, PUFA are also susceptible targets for 1 O2. The oxidizability of PUFA by 1 O2 will be proportional to the double bound index (DBI). The DBI is the average number of double bonds per fatty acid chain in the cell (14). Thus, a logical hypothesis is that cells with a higher DBI should be more sensitive to 1 O2 phototoxicity compared to cells with low PUFA content.

The goal of this study was to investigate the toxicity of 1 O2 in various tumor cells and to determine the influence of the different PUFA and protein content of the cells on lipid-derived radical formation and membrane permeability. To examine this, nine different leukemia cells were characterized for protein and lipid content and subjected to Photofrin photosensitization. Lipid-derived radical formation was detected by electron paramagnetic resonance (EPR) spin trapping with α-(4-pyridyl-1-oxide)-N-t-butyl nitronate (POBN) and membrane permeability was determined by trypan blue dye exclusion.

†Abbreviations: ATCC, American Type Culture Collection; DBI, double bond index, the average number of double bonds per fatty acid chain in a cell; DHA, docosahexaenoic acid; EPR, electron paramagnetic resonance; FBS, fetal bovine serum; 10% FBS-

RPMI, RPMI-1640 medium with 10% fetal bovine serum; FSC, forward scatter; GPx glutathione peroxidase-1; 1 O2, singlet oxygen; PBS, phosphate-buffered saline; PDT, photodynamic therapy; POBN, α-(4-pyridyl-1-oxide)-N-t-butyl nitronate; PUFA, polyunsaturated fatty acids; SSC, orthogonal light scatter.

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Table 1. Size of nine leukemia cell lines

<table>
<thead>
<tr>
<th>Cell line</th>
<th>Median diameter (μm)</th>
<th>SD‡+</th>
<th>Range* (μm)</th>
<th>Range‡§</th>
</tr>
</thead>
<tbody>
<tr>
<td>CEM</td>
<td>9§</td>
<td>1.8</td>
<td>5.2–14.1</td>
<td>8–12</td>
</tr>
<tr>
<td>HEL</td>
<td>12.1</td>
<td>1.3</td>
<td>9.1–15.8</td>
<td>10–18</td>
</tr>
<tr>
<td>HL-60</td>
<td>10§</td>
<td>1.4</td>
<td>6.1–13.6</td>
<td>8–12</td>
</tr>
<tr>
<td>K-562</td>
<td>14.3*</td>
<td>1.6</td>
<td>9.3–18.7</td>
<td>10–20</td>
</tr>
<tr>
<td>KG-1</td>
<td>10.3*</td>
<td>1.4</td>
<td>6.1–13.3</td>
<td>8–18</td>
</tr>
<tr>
<td>L1210</td>
<td>10.4*</td>
<td>1.1</td>
<td>7.5–13.3</td>
<td>8–10</td>
</tr>
<tr>
<td>Molt-4</td>
<td>11.4*</td>
<td>1.3</td>
<td>8.5–15.5</td>
<td>10–12</td>
</tr>
<tr>
<td>THP-1</td>
<td>12.5*</td>
<td>1.4</td>
<td>9.2–16.6</td>
<td>8–12</td>
</tr>
<tr>
<td>U-937</td>
<td>13.0§</td>
<td>1.8</td>
<td>9.1–18.3</td>
<td>8–18</td>
</tr>
</tbody>
</table>

*These values were determined from the FSC measured by flow cytometry, using the equation: cell diameter = (0.0893 · FSC – 0.677) μm.
†The standard deviation (SD) of the cell diameter distribution was determined from the gaussian FSC distribution. Using the equation above (FSCmedian ± SD(ge) was converted to cell diameters.
‡These values were determined using a micrometer and conventional light microscopy.
§These diameters were determined using conventional light microscopy and were used as standards for the flow cytometric determinations.

Phosphate-buffered saline (PBS), 210 mg potassium dihydrogen phosphate, 407 mg sodium monohydrogen phosphate and 9 g sodium chloride (0.9%) were dissolved in 1 L water and adjusted to pH 7.4 with 1 N HCl. Both the sodium chloride and the PBS solution were sterile filtered and stored over chelating resin (Sigma) to minimize the level of adventitious transition metals (15). α-(4-Pyridyl-1-oxide)-N-butanilnitrone (Aldrich) was prepared as a 1 M stock solution in distilled water. Ascorbic acid (10 mM) and Fe(SO₄)·7H₂O (1 mM) stock solutions in distilled water were prepared just before each experiment. Scintiguest was from Fisher.

Cell culture. Leukemia cells were acquired from the American Type Culture Collection (ATCC). HL-60 stock cells were grown in medium consisting of RPMI-1640 medium (Gibco) and 10% fetal bovine serum (FBS-RPMI) supplemented with l-glutamine (2 mM); THP-1 and KG-1 stock cells required the addition of 2-mercaptoethanol to 10% FBS-RPMI medium. All other leukemia cells were sterile filtered and stored over chelating resin (Sigma) to minimize pH 7.4 with 1 M stock solution in distilled water. Ascorbic acid (10 mM) and Fe(SO₄)·7H₂O (1 mM) stock solutions in distilled water were prepared just before each experiment. Scintiguest was from Fisher.

Cell culture. Leukemia cells were acquired from the American Type Culture Collection (ATCC). HL-60 stock cells were grown in medium consisting of RPMI-1640 medium (Gibco) and 10% fetal bovine serum (FBS-RPMI) supplemented with l-glutamine (2 mM); THP-1 and KG-1 stock cells required the addition of 2-mercaptoethanol to 10% FBS-RPMI medium. All other leukemia cells were grown in 10% FBS-RPMI medium. The HL-60, THP1 and KG1 cells were grown in 10% FBS-RPMI medium for at least 3 days before the experiment. Experiments were done in exponential growth phase.

Cell size. Unstained cells (1 × 10⁶ cells/mL RPMI medium) were imaged on a Leitz Diaplan microscope in the phase-contrast mode. Images were captured with an Optronics DEI-750 charge-coupled device camera using Matrox Inspector software. Cell diameter was measured using a Leitz stage micrometer (2 mm, intervals 0.01 mm). At least three images were captured and diameters of all cells (~5–10 cells/image) determined.

For flow cytometry measurements, cells were resuspended in 10% FBS-RPMI medium and aliquoted into tubes. Flow cytometry was performed with a Becton Dickinson FACScan (Becton Dickinson Immunocytometry Systems, San Jose, CA). CellQuest V3.1 software (Becton Dickinson) was used for acquisition and analysis of data. Forward scatter (FSC) and orthogonal scatter (SSC) were collected using linear amplification. Twenty thousand events were collected in listmode. A bitmap gate was placed around the cell population on the basis of FSC and SSC to eliminate debris and aggregates. Statistics were generated on single-parameter histograms of forward scatter after elimination of debris and aggregates using the bitmap gate.

The diameter range of the nine leukemia cell lines was determined with a phase-contrast light microscope using a micrometer as a scale reference. The diameters determined by light microscopy for all cell lines ranged from 8 to 20 μm (Table 1). While some cell lines have a narrow range of diameters, only 8–12 μm (CEM, HL-60, L1210 and Molt-4), others spanned from 8 to 20 μm (K-562 and HEL). A true statistically representative distribution of cell diameter is very cumbersome and not realistically achievable with light microscopy. To determine a statistically valid diameter distribution for these nine cell lines, FSC was measured. Polystyrene microspheres (10–20 μm) cannot be used as standards to bracket the FSC of the cells because of their very different scattering properties. Thus, to determine the diameter and the diameter distribution of the nine cell lines, we chose three cell lines that had a relatively narrow range and that approached the diameter range of all 9 cell lines (CEM, HEL, U-937) were measured by conventional light microscopy. The median diameters from light microscopy of CEM, HL-60 and U-937 cells were used as standards for the calibration curve of the FSC data (Fig. 1). The results of conventional light microscopy and the FSC of the flow cytometer were very comparable, and the standard curve has a regression coefficient of 0.999.

The diameters of the remaining six cell lines were determined using the equation of the standard curve (y = 0.0893x – 0.677, where y = cell diameter in μm, and x = FSC) (Table 1). The median diameter varies from 9–14 μm for the nine cell lines. The range was determined using the upper and lower 1% of the gaussian FSC distribution on both sides and correlates well with the range determined by conventional light microscopy.

Photofrin uptake. Cells were exposed to Photofrin (0, 0.9 or 9 μg/mL) at 37°C for 45 min in PBS (pH 7.4). After incubation with Photofrin, cells were washed by centrifugation (300 g) and resuspended in PBS, pH 7.4, or 0.9% NaCl. For the determination of Photofrin uptake (16), Photofrin-exposed cells were resuspended in 2.7 mL PBS and 300 μL Scintiguest. Scintiguest digests the cells and disaggregates the Photofrin (17). This allows a more accurate assessment of Photofrin uptake because monomers have different spectroscopic properties than dimers and oligomers. Standards and standards (various Photofrin concentrations in PBS) were heated at 60°C in the dark for 24 h. Photofrin was determined with a Perkin-Elmer LS50B fluorometer; excitation 402 nm and emission 620 nm. The cells not exposed to Photofrin were used as the blank control. Photofrin concentrations ranging from 5 ng to 2.5 μg/mL in buffer were used to construct a standard curve. Standards were treated identically to the cell samples, i.e., Scintiguest and heat. Fluorescence quenching occurs above 1 μg/g/mL Photofrin. Thus, all cell samples were diluted as necessary so that the fluorescence intensity was in the linear range of the standard curve. Protein was analyzed using the Lowry assay (18).

DBI and total lipid determination. The DBI and total lipid of all cell lines were determined following Wagner et al. (14). Briefly, cells (2 × 10⁶) were washed and extracted with chloroform:methanol 2:1 (vol/vol). After alkaline hydrolysis, fatty acids in the saponifiable fraction were methylated and the methylesters analyzed by gas–liquid chromatography.

Membrane permeability. Trypan blue dye exclusion was used to determine membrane permeability. Cells (4 × 10⁶/mL) were ex-
posed to Photofrin for 45 min, washed and resuspended in 1 mL PBS. Ascorbic acid (100 μM) and FeSO₄ (5 μM) were added to 500 μL cell sample (2 × 10⁶ cells) to catalyze radical production from lipid hydroperoxides and placed into one well of a 48 well dish. The dish was placed onto a light box (5 J/m² s) and cells exposed to light up to 32 min. Twenty microliter aliquots were removed every 2 min, mixed with 80 μL trypsin blue solution (0.2 %), and cells with membrane damage were counted under a microscope. The light box used for this experiment was equipped with a circular Sylvania light bulb (daylight, 22 W) covered with a light diffuser that is impermeable to UV. The intensity was measured with a Yellow Spring Instrument 65 A Radiometer with the 6551 probe. All samples were places in the same marked positions on the light box that provided an even fluence rate.

**Fatty acid modification.** K-562, U-937 and CEM cells were grown in 10% FBS-RPMI medium supplemented with 32 μM docosahexaenoic acid, 22:6ω₃ (DHA) for 48 h. Cells were washed twice first in RPMI-1640 medium and then in PBS by centrifugation (300 g). Upon resuspension, cell density was adjusted for trypan blue (4 × 10⁶ cells/mL) or EPR experiments (8 × 10⁶ cells/mL).

**EPR experiments.** Cells (8 × 10⁶/mL) were incubated with Photofrin for 45 min, washed and resuspended in 1 mL 0.9% NaCl. An aliquot of 500 μL was mixed with POBN (25 mM), ascorbic acid (100 μM) and FeSO₄ (5 μM) and placed into a TM EPR quartz flat cell. Ferrous iron and ascorbic acid was needed to initiate radical formation from lipid hydroperoxides. Cells were exposed to visible light (tungsten, 180 J/m² s), and the POBN radical adduct EPR signal was monitored for 32 min. Twenty microliter aliquots were removed every 2 min, mixed with 80 μL trypsin blue solution (0.2 %), and cells with membrane damage were counted under a microscope. The light box used for this experiment was equipped with a circular Sylvania light bulb (daylight, 22 W) covered with a light diffuser that is impermeable to UV. The intensity was measured with a Yellow Spring Instrument 65 A Radiometer with the 6551 probe. All samples were places in the same marked positions on the light box that provided an even fluence rate.

**RESULTS AND DISCUSSION**

**The nine leukemia cell lines studied**

In this study, one murine and eight human leukemia cell lines were used. They can be divided into different lineages (Scheme 1). CEM, Molt-4 and L1210 (murine) cells represent the lymphoid lineage. They are T-lymphoblast-like cells. The K-562 and HEL cells are erythroid cells. The K-562 cells can be differentiated into progenitors of erythrocytes, granulocytes and monocytes. The HL-60 is a promyelocytic cell line. Vitamin D₃ can differentiate the HL-60 cells beyond the promyelocytic stage into macrophages and monocytes. The KG-1 cells resemble myelogenous cells with a predominance of myeloblasts and promyelocytes. A small percentage of the cells are mature granulocytes; macrophages and eosinophils can be present. The THP-1 and U-937 are monocyte-like cells. The THP-1 cells can be differentiated into macrophage-like cells and U-937 can be induced to terminal monocytic differentiation by vitamin D₃ and other compounds.

**Changes in membrane permeability with Photofrin photosensitization are cell line dependent**

Because of the hydrophobic nature of Photofrin, it locates in cell membranes (19,20). Upon light exposure, it produces ¹⁰₂, which can oxidize polysaturated lipids forming lipid hydroperoxides (LOOH). This oxidative damage can change the membrane permeability that can be detected by trypan blue dye uptake. Cells, whose membranes are intact, exclude trypan blue; when the membrane is damaged, trypan blue can enter the cell. The membranes of different cell lines can vary greatly in their composition and with this in their susceptibility to oxidation. As shown in Fig. 2, membrane permeability after Photofrin photosensitization is cell line dependent. The t₅₀%, the time when 50% of the cells exclude trypan blue, ranges from 11.8 min (CEM cells) to 28.7 min (K-562 cells) (Table 2). Interestingly, t₅₀% of the lymphoid cells (CEM, MOLT-4 and L1210) are ≥20 min, while the t₅₀% of the myelomonocytic and erythroid cell lines are >20 min (with the exception of the HL-60 cells, 13 min). Various factors that could produce this difference in membrane damage are examined below.

**Photofrin uptake in leukemia cells is dose and cell line dependent**

Photofrin uptake is dependent on the extracellular environment (e.g. PBS, medium, medium + FBS), on pH and on time of exposure (17,21,22). To compare Photofrin phototoxicity in different cells, all nine cell lines were exposed under the same conditions and Photofrin uptake was deter-
The total lipid content of all nine cell lines varies greatly (Table 3). The CEM cells, the smallest cell line, have the lowest lipid content (4 µg/10^6 cells); K-562, the biggest cell line, has the highest lipid content (18 µg/10^6 cells). Although lipids can be a target for singlet oxygen, here they seem to protect the cell against ¹O₂ damage. Resistance to Photofrin phototoxicity (t_50%) increases with increasing lipid content (Fig. 3). There is a strong positive correlation between t_50% and total lipid (R = 0.94) (Fig. 3A); the correlation for the lymphoid cells is ideal (R = 0.999) (Fig. 3C). The correlation for the four cell lines with the same Photofrin content per cell is R = 0.993 (Fig. 3B). These data suggest that lipids can protect the plasma membrane from ¹O₂ damage.

Membrane permeability weakly correlates with DBI

The lipids of a cell contain both saturated and unsaturated fatty acids. Singlet oxygen can add to double bonds of fatty acid chains to form hydroperoxides. These hydroperoxides can undergo radical chain reactions in the presence of metals, leading to fragmentation of the lipids and damage to the cell membranes.

Table 3. Double bond index, total lipid, protein and cell size*

<table>
<thead>
<tr>
<th>Cell line</th>
<th>Total lipid (µg/10^6 cells)</th>
<th>DBI‡</th>
<th>Protein (µg/10^6 cells)</th>
<th>Cell volume§ (µm³)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CEM</td>
<td>4 ± 0.4</td>
<td>0.47 ± 0.10</td>
<td>45 ± 7</td>
<td>200</td>
</tr>
<tr>
<td>HEL</td>
<td>14 ± 1.2</td>
<td>0.62 ± 0.02</td>
<td>177 ± 1</td>
<td>486</td>
</tr>
<tr>
<td>Molt-4</td>
<td>6 ± 0.6</td>
<td>0.63 ± 0.03</td>
<td>107 ± 15</td>
<td>406</td>
</tr>
<tr>
<td>L1210</td>
<td>8 ± 0.5</td>
<td>0.73 ± 0.10</td>
<td>170 ± 15</td>
<td>308</td>
</tr>
<tr>
<td>K-562</td>
<td>18 ± 1</td>
<td>0.76 ± 0.04</td>
<td>299 ± 12</td>
<td>802</td>
</tr>
<tr>
<td>THP-1</td>
<td>12 ± 0.2</td>
<td>0.79 ± 0.01</td>
<td>112 ± 18</td>
<td>535</td>
</tr>
<tr>
<td>U-937</td>
<td>13 ± 1</td>
<td>0.95 ± 0.02</td>
<td>134 ± 45</td>
<td>602</td>
</tr>
<tr>
<td>KG-1</td>
<td>13 ± 0.2</td>
<td>1.21 ± 0.01</td>
<td>75 ± 5</td>
<td>300</td>
</tr>
<tr>
<td>HL-60</td>
<td>9 ± 0.3</td>
<td>1.23 ± 0.07</td>
<td>122 ± 7</td>
<td>277</td>
</tr>
<tr>
<td>K-562 DHA enriched</td>
<td>28 ± 3.3</td>
<td>2.38 ± 0.08</td>
<td></td>
<td></td>
</tr>
<tr>
<td>U-937 DHA enriched</td>
<td>22 ± 1.1</td>
<td>2.91 ± 0.06</td>
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</tbody>
</table>

*Results are the mean of at least three independent experiments.
‡DBI, the average number of double bonds per fatty acid chain in a lipid ensemble (such as a cell membrane).
§Cell volume was calculated assuming cells are spherical. Numbers represent the median volume.
Figure 3. Correlation of total lipid and membrane permeability in leukemia cells after exposure to Photofrin and light. The t50% is the time when 50% of the cells exclude trypan blue during Photofrin and light exposure. (A) All nine leukemia cell lines; (B) CEM, K-562, Molt-4, and U-937 cells (they have comparable amount of Photofrin); (C) lymphoid cell lines (CEM, L1210 and Molt-4); (D) myelomonocytic and erythroid cell lines. The data are derived from Fig. 2 and Table 3.

Figure 4. Membrane permeability in leukemia cells after exposure to Photofrin and light is not influenced by lipid unsaturation. The t50% is the time when 50% of the cells exclude trypan blue during Photofrin and light exposure. (A) All nine leukemia cell lines (data are derived from Fig. 2 and Table 3); (B) CEM, K-562, Molt-4 and U-937 cells (they have the same amount of Photofrin/cell); (C) K-562 cells enriched with DHA for 48 h prior to Photofrin (0.9 μg/mL for 45 min in PBS) photosensitization; (D) U-937 cells enriched with DHA for 48 h prior to Photofrin (0.9 μg/mL for 45 min in PBS) photosensitization. Experiments were done in triplicate.
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Figure 5. Protein protects leukemia cells from Photofrin photosensitization. The t 50% is the time when 50% of the cells exclude trypan blue during Photofrin and light exposure. (A) All nine cell lines; (B) CEM, K-562, Molt-4 and U-937 cells (they have comparable amount of Photofrin); (C) lymphoid cell lines; (D) myelomonocytic and erythroid cell lines. The data derived from Fig. 2 and Table 3.

Protein protects leukemia cells from Photofrin photosensitization

It has been demonstrated that 1 O₂ produced in a cell membrane readily diffuses into the neighboring aqueous environments inside and outside the cell (23). Singlet oxygen can be quenched by water, lipids, proteins, nucleic acids and other small molecules. When 1 O₂ is produced in a cell membrane, up to 80% can be quenched by protein; unsaturated lipids can account for up to 7% of the quenching, depending on cell type (12,13). The quenching of 1 O₂ by unsaturated lipids is dominated by chemical processes that result in the formation of lipid hydroperoxides that cause membrane damage; quenching by protein can be both chemical and physical, depending on the amino acids involved.

The protein content and cell size of the nine cell lines vary greatly. Protein content ranges from 45 μg/10⁶ for CEM cells to 299 μg/10⁶ for K-562 cells; the cell diameter ranges from 9 μm (CEM) to 14 μm (K-562) (Table 1). Resistance to Photofrin phototoxicity (t₅₀%) increases with increasing protein (Fig. 5A). These observations are consistent with the known ability of some amino acids to quench singlet oxygen (8–11). Chemical quenching reactions, such as occur with cysteine, tryptophan and methionine, could be detrimental, while physical quenching, such as with histidine, would offer protection. The myelomonocytic and erythroid cells show only a weak correlation (R = 0.49) between t₅₀% and protein content (Fig. 5D).

In contrast, the remarkable positive correlation (R = 0.999, Fig. 5C) of t₅₀% for the lymphoid cell lines with cellular protein levels and the correlation (R = 0.937, Fig. 5B) for the four cell lines that have the same amount of Photofrin per cell suggest that the membrane-associated proteins of these cells offer a unique mode of protection. The K-562 cells have the highest protein to DBI ratio (393 μg protein/10⁶ cells : DBI) and also the highest t₅₀%, while CEM and HL-60 cells each have a low ratio (95 μg protein/10⁶ cells : DBI) and a correspondingly low t₅₀%. Therefore a high ratio of protein to unsaturated lipid in a cell might be important for protection against 1 O₂. Lipids and proteins provide a competing environment for 1 O₂; different cell lines present different environments and respond accordingly.

Membrane susceptibility to photodamage by Photofrin decreases with increasing cell size

The size of the nine leukemia cell lines measured as volume (Table 3) ranges from 200 μm³ (CEM) to 800 μm³ (K-562). Resistance to Photofrin phototoxicity (t₅₀%) increases with
Figure 6. Membranes of bigger cells are less sensitive to Photofrin photosensitization. The $t_{50\%}$ is the time when 50% of the cells exclude trypan blue during Photofrin and light exposure. (A) Correlation between membrane permeability and cell size; (B) correlation between total lipid content of cells and cell size; (C) correlation between cell protein and cell size; (D) correlation between DBI and cell size for all nine leukemia cells; (E) correlation between DBI and cell size for myelomonocytic and erythroid cells; (F) correlation between Photofrin uptake and cell size. The data derived from Fig. 2 and Tables 2 and 3.

Increasing cell volume (Fig. 6A). Variants of RIF-1 cells with larger diameters showed a tendency to increase resistance, but unfortunately actual cell volumes were not determined (24). With increasing cell volume, the total lipid and total protein content naturally increases (Fig. 6B,C). In contrast to the total lipid, DBI for all nine cell lines shows no change with volume (Fig. 6D). However, when the subset consisting of only the myelomonocytic and erythroid cell lines is examined, a decrease in the DBI with increasing volume is seen (Fig. 6E). Photofrin uptake does not change with increasing volume (Fig. 6F). Thus bigger cells compared to smaller cells should produce the same amount of $1O_2$ because they have comparable amounts of Photofrin. However, bigger cells are less damaged because the levels of proteins and total lipids are higher. While the lipid content of larger cells is greater, the degree of unsaturation is less (Fig. 6E) making them less susceptible to lipid peroxidation. Thus, membrane damage induced by Photofrin photosensitization is a function of lipid, protein and Photofrin content of the cell.

Radical formation increases with increasing DBI

Singlet oxygen reacts with unsaturated lipids forming lipid hydroperoxides (LOOH). These hydroperoxides can react with ferrous iron initiating free radical chain reactions producing oxygen- ($LO_2\;OLOO_2\;O$) and carbon- ($L_d$) centered radicals. The rate of these oxidations should increase with increasing DBI of the cells. Using EPR with POBN as a spin trap, we detected carbon-centered radicals formed as a result of the lipid peroxidation processes initiated by Photofrin photosensitization (Fig. 7A). This radical formation increases with duration of light exposure and reaches a plateau for most cell lines at approximately 12 min. Radical formation is cell line dependent; HL-60 cells produce the highest flux and HEL cells the lowest. No correlation between radical formation and total lipid was found (Fig. 8A); when cell lines with similar levels of Photofrin are compared, DBI correlates well with radical formation ($R = 0.946$) (Fig. 8C).

The DBI of cells can be increased by supplementing the growth media with linoleic acid (18:2) and highly unsatu-
Linoleic acid (K18:2)- and DHA (K22:6)-enriched K-562 cells treat-
bars are only shown for HL-60 and HEL for clarity of figure. (B) the sample was then illuminated with visible light (180 J/m² s) in
the radical flux observed can derive from all cell membranes
by the amount of unsaturated lipids in the plasma membrane,
(Fig. 4C,D). While the membrane permeability is influenced
that was not influenced by DHA enrichment in K-562 cells
¯ux (Fig. 7B). This is in contrast to membrane permeability
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in DBI is not high enough in linoleic acid-enriched cells to
0.04 to 0.92
6
6
linoleic acid results in an increase in the DBI from 0.76
6
6
DBI from 0.76
0.08 (Table 3). The increase
in greater radical formation. Enrichment of K-562 cells with
rated DHA (22:6). This increase in unsaturation should result
in greater radical formation. Enrichment of K-562 cells with
linoleic acid results in an increase in the DBI from 0.76 ±
0.04 to 0.92 ± 0.01 while DHA results in an increase in the
DBI from 0.76 ± 0.05 to 2.38 ± 0.08 (Table 3). The increase
in DBI is not high enough in linoleic acid-enriched cells to
have a detectable effect on the radical flux. However, DHA-
enriched cells produced an increase in the observed radical
flux (Fig. 7B). This is in contrast to membrane permeability
that was not influenced by DHA enrichment in K-562 cells
(Fig. 4C,D). While the membrane permeability is influenced
by the amount of unsaturated lipids in the plasma membrane,
the radical flux observed can derive from all cell membranes
that bind Photofrin (mitochondria, plasma membrane, en-
doplasmic reticulum) and the neutral lipids stored in droplets
in the cytosol. These results indicate that unsaturated lipids
are a target for the ¹O₂ that is produced during Photofrin
photosensitization.

Protein does not easily undergo oxidation by radical chain
reactions. Indeed, radical formation does not increase with
increasing protein levels (Fig. 8D) or increasing volume
(Fig. 8E). These results are comparable with the membrane
permeability. The proteins appear to quench mainly physically
because no protein radicals were detected.

Thus, radical formation induced by ¹O₂ appears to be a
function of lipid content of the cell. Highly unsaturated lip-
ids increase radical formation.

CONCLUSIONS

The oxidative damage that results from ¹O₂ exposure is de-
pendent on cell line. Different cell lines present different
environments and respond accordingly. Using Photofrin as
a source of ¹O₂, we examined various factors in cells that
influence the toxicity of ¹O₂.

Photofrin. We observed that cell lines with comparable
amounts of Photofrin/cell had different susceptibility to dam-
age by ¹O₂.

Lipids. Saturated lipids are not a target for ¹O₂ while un-
saturated lipids can be easily oxidized. Oxidation of PUFA
leads to LOOH formation. If no free redox-active metal is
available, these LOOH can accumulate. The LOOH can be
removed by phospholipase A₁/glutathione peroxidase-1
(GPx) or phospholipid hydroperoxide GPx (PhGPx) (25). If
ferrous iron is available, it can react with LOOH initiating
free radical chain reactions that would be detrimental to the
cell. We detected lipid-derived radical formation in all nine
leukemia cell lines exposed to Photofrin and light. Increasing
the unsaturation of the cell lipids increases the radical for-
amation. Surprisingly we observed no correlation between
lipid unsaturation (DBI) and change in membrane perme-
ability (t⁵₀) after ¹O₂ exposure. Kanofsky (13) has shown
that only approximately 7% of the ¹O₂ produced in mem-
branes is quenched by cellular lipids. This is in agreement
with our data that suggest that lipids may not be the primary
target for ¹O₂-mediated phototoxicity.

Protein. Proteins appear to quench about 80% of the ¹O₂
produced in cell membranes (13). Singlet oxygen can be
physically quenched by proteins, which results in protection
against cell damage. Proteins can also chemically quench
¹O₂. The chemical quenching leads to damage of the protein
that can bring about membrane depolarization (26,27) and if
severe be lethal. The typical rate constants reported for un-
saturated lipids reacting with ¹O₂ in various organic solvents
is on the order of 10³ M⁻¹ s⁻¹ (8). In contrast, rate constants
for ¹O₂ reacting with proteins are considerably greater. For
example: histidine k > 10⁸ M⁻¹ s⁻¹; tryptophan k > 10⁹ M⁻¹
s⁻¹; cysteine k = 10⁶ M⁻¹ s⁻¹; albumin k ∼ 2–5 × 10⁹ M⁻¹
s⁻¹; Cu and Zn containing superoxide dismutase k = 10⁸
M⁻¹ s⁻¹. We found an inverse correlation between membrane
permeability changes (t⁵₀) and cellular protein suggesting
that proteins play an important role in protection against ¹O₂
toxicity.

Cell size. A remarkable observation is that larger cells are
more protected against Photofrin phototoxicity than smaller cells. Under our experimental conditions, the amount of Photofrin uptake was independent of cell size, but as expected the protein and lipid content increases with increasing size. This increase in lipid and protein content provides a larger capacity to quench \( ^1\text{O}_2 \). Thus, larger cells are more resistant to a given flux of \( ^1\text{O}_2 \).

If \( ^1\text{O}_2 \) is generated in tissue, the lipid and protein profile of that tissue will strongly influence the extent of damage produced.

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